207. Experiments on the Total Synthesis of Lysolipin I

Part III¹)

Preparation and Transformations of Substituted 1,2,3,4-Tetrahydrodibenzofuran-1-ones²)

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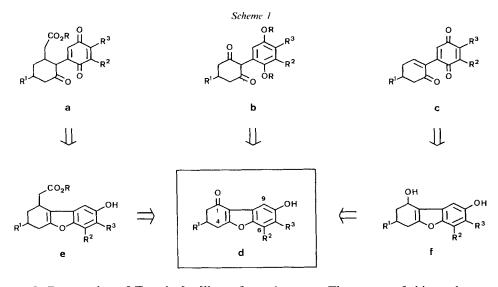
Summary

1,2,3,4-Tetrahydrodibenzofuran-1-ones were obtained by *Michael* addition of 1,3cyclohexadione (2) to o-benzoquinone (3) and to p-benzoquinones 8 and 11 (Scheme 2). In addition to the expected 7,8-disubstituted adduct 14, the ZnCl₂-catalyzed reaction of dione 2 with methoxy-p-benzoquinone (11) afforded a small amount of the 6,8-disubstituted regio-isomer 13 (Scheme 2). The projected cleavage of these dibenzofuranones to 3-methoxy-2-phenyl-2-cyclohexenone 22 could be effected by treatment with NaOH followed by methylation (Scheme 3). Attempted acetalization of such dibenzofuranones resulted in a retro-Claisen-type cleavage, giving the benzofuryl-butyrate 16. Other transformations include reduction of the ketone, of the C(4a)=C(9b) bond, and alkylation with Li-ethoxyacetylide (Scheme 3). Oxidation of 8-hydroxy-7methoxydibenzofuran derivatives led to o-quinones instead of the desired ring cleavage to p-quinones (Scheme 4).

1. Introduction. – A retrosynthetic analysis of the complex structure determined for the antibiotic Lysolipin I [3] led to (3'-oxocyclohexyl)acetic acids with a p-hydroquinoid substituent at C(2') as starting point for a promising approach to this synthetic problem (cf. [1] [4]). Such structural units should readily be available from either the corresponding quinones **a**, cyclohexadiones **b**, or cyclohexenones **c**, which in turn are derivable from 1,2,3,4-tetrahydrodibenzofuran-1-ones **d** by hydrolytic (\rightarrow **b**) or oxidative (\rightarrow **a** via **e**, \rightarrow **c** via **f**) opening of the furan ring (Scheme 1). 1,2,3,4-Tetrahydrodibenzofuran-1-ones have been prepared by several methods [1] [5] [6], and the projected hydrolytic cleavage of the furan ring (**d** \rightarrow **b**) was successful with the closely related 3-acyl-2-alkyl-benzofurans [7]. 2-Hydroxydibenzofuran, on the other hand, was cleaved to (2'-hydroxyphenyl)-p-benzoquinone by periodate oxidation [8].

¹) Part II [1].

²) Part of these results are included in the Ph. D. thesis of V. Sch. [2].



2. Preparation of Tetrahydrodibenzofuran-1-ones. – The outset of this study was given by the facile access to dibenzofuranone 1, which is formed from 1,3-cyclohexadione (2) and o-benzoquinone (3), obtained by *in situ* oxidation of catechol (4) [5]. Alkylation of 1 with $(CH_3O)_2SO_2/K_2CO_3$ gave the dimethoxy compound 5 in 22% overall yield (Scheme 2)³). For two reasons we looked for alternative approaches to such dibenzofuranones. 1) The planned oxidations $(\mathbf{f} \rightarrow \mathbf{c}, \mathbf{e} \rightarrow \mathbf{a}, Scheme 1)$ require selectively O-protected derivatives of 1 to prevent oxidation to o-quinones (partial O-methylation of 1 afforded a 1:1-mixture of 6 and 7 [2], Scheme 2). 2) Although the addition to o-benzoquinone (3) can be extended to 5-substituted 1,3-cyclohexadiones [5], 6-substituted dibenzofuranones of type d (Scheme 1) are not necessarily available by this method.

Since malonic-acid derivatives and acetoacetates have been found to add readily to p-benzoquinones under acid [9] or base [10] catalysis, and since the regioselectivity of these *Michael* additions is well-understood and predictable [11], we decided to investigate the 1,4-addition of cyclohexadione **2** to p-benzoquinones.

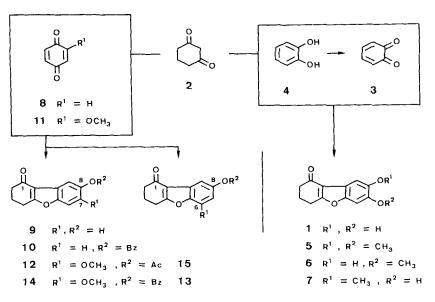
Reaction of *p*-benzoquinone (8) with a twofold excess of 2 in *t*-BuOH/Et₂O (or *t*-BuOH/CH₂Cl₂)⁴) catalyzed by ZnCl₂ according to [9a] followed by treatment with CH₃OH/CH₃SO₃H/ClCH₂CH₂Cl⁵) and benzylation of the primary product 9 gave the dibenzofuranone 10 in very low yield $(10\%)^6$). The 7-methoxy-substituted derivative 12 (isolated in 24% yield by acetylation) was obtained analogously from methoxy-*p*-ben-

³) Much higher yields of 1 (>90%) have been described in the literature [5b] [5c]. The success of this transformation depends on the rates of oxidation and removal from the reaction medium of the product which is itself sensitive to oxidation.

⁴) In CH₃OH or EtOH etherification of dione **2** by the solvent was a major side-reaction, while in aprotic media (Et₂O, acetone) redox processes were found to predominate.

⁵) Under these conditions [12] excess dione 2 is converted to its methyl-enol ether, which has advantageous properties in chromatographic separations.

⁶) The acetate of dibenzofuranone 9 was obtained much more efficiently (77% yield) by using p-benzoquinone monoethylene acetal [1].



zoquinone $(11)^7$). Closer examination of the mixture obtained by ZnCl₂-catalyzed addition of dione 2 to quinone 11 followed by benzylation led to the detection of the regio-isomer 13, which was isolated in *ca*. 6% yield in addition to 25% of 14 (*Scheme 2*)⁸).

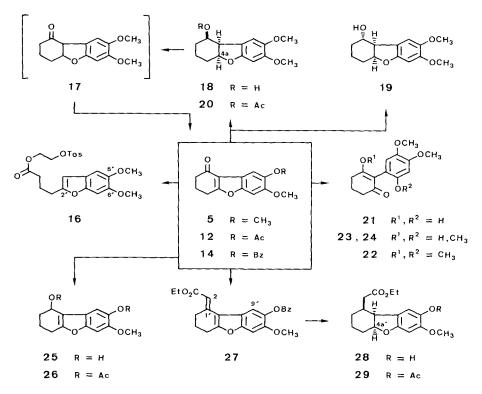
The unsatisfactory yields of these reactions are in part due to the oxidation of the primary *Michael* adducts (possibly before furan-ring closure) by the starting quinones 8 and 11, respectively, leading to complex products and the hydroquinone of the starting material. Interestingly, the TiCl₄-catalyzed reaction of the silylenol ether derived from 2 with quinone 11 gave a high yield of 2-methoxyhydroquinone [2], indicating a direct redox process between the reactants. Similar observations have been made with quinone monoacetals [1].

3. Transformations of 1,2,3,4-Tetrahydrodibenzofuran-1-ones. – The hydrolytic opening of the furan ring of 1,2,3,4-tetrahydrodibenzofuran-1-ones proved to be more difficult than anticipated. Dimethoxy compound 5 and the analogous compound obtained from dimedone and catechol (4) (cf. [2] [5]) resisted treatment with 10% aq. KOH in dioxane in the presence of $(CH_3O)_2SO_2$ at reflux temperature, KOCH₃ in boiling Et₂O or CH₃OH, followed by addition of $(CH_3O)_2SO_2$, trimethyl orthoformate/ CH₃SO₃H in boiling benzene, or CH₃OH/HCl at 70° (cf. [2]). Under acetalization conditions (1,2-ethanediol/TsOH·H₂O) ketone 5 was transformed to 4-(2'-benzofuryl)butyrate 16 (Scheme 3). To prepare the 4a,9b-dihydro-derivative 17, 5 was hydrogenated at 3-5 atm. pressure with 10% Pd/BaSO₄ in AcOEt, affording the hexahydro-dibenzofuran-1-ols 18 and 19 in a ca. 3:1 ratio. Alcohol 18 was also characterized as its

⁷) Benzofuranone 12 was also obtained, albeit in even lower yield, by addition of the morpholino-enamine derived from dione 2 to quinone 11 according to [10c].

⁸) While separation of the acetates 12 and 15 is much more difficult than the separation of the benzyl ethers 13 and 14, the 8-(2'-tetrahydropyranyloxy) derivatives could also be separated easily by chromatography. Acetate 15 was obtained pure by transformation of the corresponding tetrahydropyranyl ether (cf. [2]).





acetate **20**⁹)¹⁰). Oxidation of **18** with a variety of reagents including DMSO/pyridine \cdot SO₃ [13] gave exclusive formation of **5** with no traces of **17** detectable (*Scheme 3*, *cf.* [2])¹¹).

It was discovered only recently, that a clean hydrolytic cleavage of 5 is possible by treatment with NaOH in aq. EtOH according to [7]. It is quite remarkable, that the 1,3-cyclohexadione moiety is unaffected under these conditions¹²), while the analogous acyclic β -diketones obtained from 3-acyl-2-alkylbenzofurans undergo quantitative *retro-Claisen* reaction [7]¹³). Quenching with CH₃I afforded a mixture of the tetramethoxy derivative **22** and partially methylated compounds **23** and **24**. Methylation of **23**

⁹) Hydrogenation with 5% Pd/C at normal pressure led to extensive hydrogenolysis of the O-function at C(1) (cf. [2]).

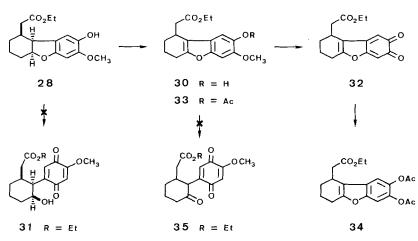
¹⁰) The structural assignment of the isomers **18** and **19** relies on the assumptions, that the furan double bond is hydrogenated in a *cis*-fashion, and that the main product **18** results from attack *trans* to OH-C(1). Hydrogenation of ketone **17** from the less hindered *exo*-face would also lead to **18**. An unambiguous corroboration of these structures with the aid of the ¹H-NMR spectra (300 MHz) was unfortunately thwarted by the rather unusual coupling pattern of **18/20** and **19** (see *Exper. Part*).

¹¹) Compound **17** might be converted to 5 by autoxidation. Such an oxidation was recently observed in the case of an analogous carbazole derivative [14].

¹²) The ketone cleavage of 1,3-cyclohexadiones requires boiling under reflux in aq. base for 30 h [15].

¹³) Careful acidification and extraction (AcOEt) allows the isolation of the polar crystalline 2-aryl-1,3-cyclohexadione 21, which is slowly reconverted to 5 (no exp. description).





and 24 (CH₃I/K₂CO₃/acetone) finally afforded 22 in a combined yield of 67% (*Scheme* 3)¹⁴). To test the possibility of an oxidative cleavage of the furan ring, the acetate 12 was reduced with NaBH₄ to the alcohol 25, characterized as diacetate 26. The acetic acid side-chain was introduced by alkylation of 14 with Li-ethoxyacetylide [16] followed by acid treatment, affording the unsaturated ester 27 in high yield¹⁵). The hexa-hydro-dibenzofurylacetate 28, characterized as acetate 29, was finally obtained from 27 by hydrogenation (*Scheme* 3)¹⁶).

Unfortunately, the oxidation of 28 with periodic acid according to [19] did not proceed as expected. Instead of ring cleavage, affording the desired *p*-benzoquinone 31^{17}), dehydrogenation to 30 and further oxidation led to the *o*-benzoquinone 32, as demonstrated by reduction of the crude mixture with dithionite and acetylation, which gave the tetrahydro-dibenzofuranylacetate 33 (10%) and *ca.* 24% of the diacetate 34 (*Scheme 4*). This result also explains the unavailability of *p*-benzoquinone 35 by oxidation of 30 (*Scheme 4*) and other fruitless attempts at the cleavage of the furan ring of 7-methoxy-8-dibenzofuranol derivatives by oxidation (*e.g.* 25, *cf.* [2]).

¹⁴) In addition to 22, 10% of starting material 5, formed in the course of the alkylation of 21, were isolated. Treatment of 21¹³) with (CH₃O)₂SO₂/K₂CO₃ in boiling acetone gave a 1:1 mixture of 22 and 5 (no exper. description).

¹⁵) Related 1,2,3,4-tetrahydrodibenzofuran-1-ones have been alkylated successfully with the *Reformatzkii* reagent derived from bromoacetate [6a] [17]. In contrast to 3-ethoxy-2-cyclohexenone [18] such dibenzofuranones have been found to be inert towards the *Wittig-Horner* reagent generated from triethyl phosphono-acetate (cf. [2]). The (E)-configuration of the C(2)=C(1') bond follows from ¹H-NMR measurements. The spatial proximity of H-C(2) and H-C(9') was deduced by a difference-NOE experiment (irradiation at the resonance frequency of H-C(2) gives enhanced sensitivity of H-C(9')).

¹⁶) The C(2)=C(1') bond is hydrogenated at a faster rate than the furane ring. Under controlled conditions the dihydro derivative 30 (Scheme 4) can therefore be isolated (cf. [2] [6a]). The relative configuration of 28 relies on similar assumptions as used for the assignment of 18 and 19¹⁰). The stereochemical relationship between 18 and 28 was indicated by their ¹H-NMR spectra exhibiting similar coupling parameters for the bridgehead H-atoms H-C(4a) (18) and H-C(4a') (28), respectively.

¹⁷) Oxidation of the related 2,4-dimethoxyphenol with periodate/AcOH gives 60% of methoxy-*p*-benzoquinone (11) [8a].

It is, thus, demonstrated that tetrahydro-dibenzofuranones **d** can be transformed hydrolytically but not by oxidation to 2-aryl-1,3-cyclohexadiones **b** (*Scheme 1*), which in turn should be transformable to (2'aryl-3'-oxocyclohexyl) acetic acids by routine procedures. Such a derivative, obtained in this special case by an alternative procedure [2] [4], could be cyclized to an octahydro-phenanthrenedione, a key compound of a projected total synthesis of Lysolipin I [4].

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Experimental Part

General. See [20].

1. Preparation of Tetrahydrodibenzofuran-1-ones. – 1.1. 7,8-Dimethoxy-1,2,3,4-tetrahydrodibenzofuranone (5). A solution of KlO₃ (1.4 g, 6.54 mmol) and NaOAc· $3H_2O$ (3 g, 22.1 mmol) in H_2O (25 ml) was added dropwise to a solution of 1,3-cyclohexadione (2) (1.12 g, 10 mmol), catechol (4) (1.1 g, 10 mmol), and NaOAc· $3H_2O$ (6 g, 44.1 mmol) in H_2O (15 ml) and acetone (3 ml). The precipitated adduct 1 (644 mg) was collected by filtration, dissolved in acetone (30 ml), and treated with (CH₃O)₂SO₂ (1.13 ml, *ca*. 12 mmol)/K₂CO₃ (1.8 g) for 5 h under reflux. Workup with AcOEt and crystallization (AcOEt/C₆H₆) afforded 540 mg (22% based on 2 or 4) of 5, m.p. 164–164.5°. IR (CHCl₃): 3100w, 3030w, 2960m, 2945m, 2915w, 2878w, 2840w, 1667s, 1627w, 1590m, 1495s, 1474s, 1467s, 1455m, 1442s, 1431m, 1418w, 1396m, 1363m, 1341w, 1320m, 1295s, 1180w, 1160m, 1140m, 1120s, 1060m, 1045m, 1020s, 1008s, 953m, 875w, 868w. ¹H-NMR (100 MHz, CDCl₃): 2.1–2.4 (*m*, 4 main peaks, 2H–C(3)); 2.46–2.66 and 2.85–3.1 (*2m*, *each 3 main peaks*, 2H–C(2), 2H–C(4)); 3.88 and 3.92 (2s. CH₃O–C(7), CH₃O–C(8)); 6.99 (*s*, H–C(6)); 7.46 (*s*, H–C(9)). MS(*di*.): 246 (100, M^{+}), 231 (55), 218 (4), 203 (15), 190 (18), 185 (7), 175 (9), 162 (7), 147 (13), 133 (7), 55 (11). Anal. calc. for C₁₄H₁₄O₄ (246.25): C 68.28, H 5.73; found: C 68.27, H 5.82.

1.2. 8-Benzyloxy-1,2,3,4-tetrahydrodibenzofuran-1-one (10). To a solution of dione 2 (6.70 g, 60 mmol) and ZnCl₂ (4.08 g, 30 mmol, dried by melting at h.v.) in t-BuOH (60 ml), a solution of p-benzoquinone (8, 3.24 g, 30 mmol) in Et₂O (150 ml) was added within 48 h at 70-80° (Ar). After workup with CH₂Cl₂, the crude material was dissolved in CH₃OH (75 ml), ClCH₂CH₂Cl (150 ml), and CH₃SO₃H(0.5 ml), and this mixture was boiled under reflux for 24 h. Workup with CH₂Cl₂, treatment with BzBr (9.5 ml, *ca.* 80 mmol)/K₂CO₃ (55.2 g) in acetone (200 ml) for 36 h (reflux), workup with CH₂Cl₂, and chromatography (200 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 6:12:1) yielded 867 mg (10% based on 8) of 10, m.p. 107–107.5° (AcOEt/cyclohexane). IR (CHCl₃): 3085w, 3060w, 2955w, 2885w, 2870w, 2840w, 1662s, 1612m, 1586s, 1495w, 1476s, 1451s, 1430w, 1399s, 1383m, 1355w, 1327w, 1317w, 1273s, 1175s, 1135m, 1135m, 1105w, 1078w, 1058m, 1043w, 1009s, 945m, 895w, 885w, 872w. ¹H-NMR (100 MHz, CDCl₃): 2.1–2.4 (m, 2H–C(3)); 2.44–2.66 and 2.80–3.10 (2m, *each 3 main signals*, 2H–C(2), 2H–C(4)); 5.08 (m, $w_{12} \approx 2$, C₆H₅CH₂O–C(8)); 6.93 (*dd*, *J* = 9 and 3, H–C(7)); 7.16–7.54 (m, H–C(6), C₆H₅CH₂O–C(8)); 7.61 (*d*, *J* = 3, H–C(9)). MS(*di*.): 292 (83, *M*⁺), 264 (6), 263 (6), 214 (11), 201 (11), 91 (100), 65 (19).

1.3. 7-Methoxy-1-oxo-1,2,3,4-tetrahydrodibenzofur-8-yl Acetate (12). To a solution of ZnCl₂ (1.36 g, 10 mmol, dried by melting at h.v.) and dione 2 (224 mg, 20 mmol) in dry EtOH (10 ml) and dry Et₂O (10 ml), heated in an oil bath of 70° (reflux), a solution of methoxy-*p*-benzoquinone (11, [10a]) (922 mg, 6.68 mmol) in EtOH (20 ml) was added within 60 h. Workup with CH₂Cl₂, acetylation (Ac₂O/pyridine), and chromatography (150 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 2:2:1) gave 170 mg (11%) of 2-methoxyphen-1,4-diyl diacetate and 448 mg (24% based on 11) of 12, m.p. 160–160.5°. IR (CHCl₃): 3035w, 2960w, 2945w, 2895w, 2875w, 2850w, 1760s, 1672s, 1637w, 1592m, 1490s, 1465w, 1453m, 1440s, 1431m, 1416w, 1397w, 1360m, 1354m, 1340w, 1330w, 1319m, 1298s, 1189s, 1175s, 1142s, 1119s, 1057w, 1043w, 1028m, 1006s, 957w, 920m, 887w, 882w. ¹H-NMR (100 MHz, CDCl₃): 2.1-2.4 (*m*, 2H--C(3)); 2.29 (*s*, CH₃COO-C(8)); 2.44-2.66 and 2.86-3.1 (2*m*, each 3 main peaks, 2H--C(2), 2H--C(4)); 3.83 (*s*, CH₃O-C(7)); 7.05 (*s*, H--C(6)); 7.67 (*s*, H--C(9)). MS(*di*.): 274 (17, M^{+}), 232 (100), 217 (50), 204 (11), 189 (5), 176 (12), 91 (3), 69 (5), 55 (4), 43 (8). Anal. calc. for C₁₅H₁₄O₅ (274.26): C 65.69, H 5.15; found; C 65.52, H 5.34.

1.4. ZnCl₂-Catalyzed Addition of Dione 2 to Quinone 11, followed by Benzylation. To a solution of 2 (6.72 g, 60 mmol) and ZnCl₂ (4.08 g, 30 mmol, dried by melting at h.v.) in t-BuOH (60 ml) a solution of 11 (4.14 g, 30

mmol, freshly sublimed) in CH₂Cl₂ (60 ml) was added within 108 h at 60–70° (exclusion of light, Ar). After heating under reflux for 20 h, part of the solvent was evaporated, and the mixture was worked up with CH₂Cl₂. The crude mixture, dissolved in CH₃OH (75 ml), ClCH₂CH₂Cl (150 ml), and CH₃SO₃H (0.5 ml), was heated under reflux for 16 h. Workup with CH₂Cl₂, treatment with BzBr (9.5 ml, *ca.* 80 mmol)/K₂CO₃ (55.2 g) in acetone (200 ml) for 60 h under reflux, workup with CH₂Cl₂ and several chromatographic separations (silica gel, cyclohexane/CH₂Cl₂/AcOEt 6:12:1) afforded 530 mg (5.5% based on 11) of 13, 2.369 g (24.5% based on 11) of pure 14, and 114 mg ($\approx 1\%$) of mixed fractions (mainly 14).

8-Benzyloxy-6-methoxy-1,2,3,4-tetrahydrodibenzofuran-1-one (13). M.p. 147–148°. IR (CHCl₃): 3030w, 2955m, 2940m, 2885w, 2835w, 1662s, 1640s, 1603s, 1587m, 1488s, 1461m, 1450m, 1440s, 1415w, 1407w, 1380m, 1357w, 1340m, 1315m, 1298s, 1174s, 1150s, 1113w, 1078w, 1055m, 1025m, 1000s, 950w, 902w. ¹H-NMR (100 MHz, CDCl₃): 2.06–2.38 (m, 2H–C(3)); 2.44–2.66 and 2.86–3.1 (2m, each 3 main signals, 2H–C(2), 2H–C(4)); 3.81 (s, CH₃O–C(6)); 5.07 (m, $w_{V_2} \approx 2$, C₆H₅CH₂O–C(8)); 6.52 (d, J = 2, H–C(7)); 7.14–7.56 (m, H–C(9), C₆H₅CH₂O–C(8)). MS(di.): 322 (100, M⁺), 245 (19), 244 (25), 231 (5), 203 (7), 91 (95). Anal. calc. for C₂₀H₁₈O₄ (322.34): C 74.52, H 5.63; found: C 74.43, H 5.65.

8-Benzyloxy-7-methoxy-1,2,3,4-tetrahydrodibenzofuran-1-one (14). M.p. 140–141°. IR (CHCl₃): 2940w, 2890w, 2870w, 2840w, 1653s, 1627w, 1586m, 1490s, 1470m, 1460m, 1450s, 1443m, 1394w, 1360m, 1340w, 1316m, 1281s, 1179w, 1155m, 1140m, 1119s, 1055w, 1043w, 1025w, 1007s, 954w, 913w, 885w, 877w. ¹H-NMR (100 MHz, CDCl₃): 2.1–2.36 (m, 2H–C(3)); 2.44–2.64 and 2.84–3.05 (2m, each 3 main peaks, 2H–C(2), 2H–C(4)); 3.86 (s, CH₃O–C(7)); 5.14 (m, $w_{Y_2} \approx 2$, C₆H₅CH₂O–C(8)); 7.0 (s, H–C(6)); 7.56 (s, H–C(9)); 7.2–7.6 (m, C₆H₅CH₂O–C(8)). MS(di.): 322 (100, M⁺), 279 (1), 231 (97), 203 (20), 91 (67), 55 (15), 43 (5).

2. Transformations of Tetrahydro-1-dibenzofuranones. – 2.1. 2-Tosyloxyethyl 4-(5',6'-Dimethoxy-2'-benzo-furyl)butyrate (16). A mixture of dibenzofuranone 5 (60 mg, 0.244 mmol), TsOH \cdot H₂O (91 mg, 0.48 mmol), and 1,2-ethanediol (0.1 ml, *ca*. 1.8 mmol) in toluene (6 ml) was boiled under reflux at a *Dean-Stark* trap for 3 h (Ar). Workup with AcOEt and chromatography (15 g of silica gel, CH₂Cl₂/AcOEt 9:1) gave 58 mg (52%, 62% based on consumed 5) of 16 and 10 mg (17%) of 5. IR (CHCl₃): 3090w, 3030w, 2960m, 2945m, 2915w, 2880w, 2840w, 1738s, 1622w, 1600m, 1484s, 1466s, 1455m, 1440s, 1400w, 1367s, 1320s, 1309m, 1293m, 1176s, 1148s, 1115s, 1097m, 1070w, 1020m, 1006m, 935m, 887m. ¹H-NMR (100 MHz, CDCl₃): 1.8–2.16 (*m*, 2H–C(3)); 2.22–2.44 and 2.6–2.8 (2*m*, *each* 3 main peaks, 2H–C(2), 2H–C(4)); 2.38, (*m*, *w*₁ ≈ 2, CH₃C₆H₄SO₃–C(2''); 3.86 (*s*, CH₃O–C(5'), CH₃O–C(6')); 3.19 (*m*, *w*₁ ≈ 2, 2H–C(1''), 2H–C(2'')); 6.28 (*m*, *w*₁ ≈ 2, H–C(3')); 6.92 and 6.98 (2*s*, H–C(4'), H–C(7')); 7.20–7.36 and 7.68–7.84 (2*m*, CH₃C₆H₄SO₃–C(2'')). MS(*di*.): 279 (33), 167 (90), 155 (7), 129 (53), 113 (37), 112 (30), 111 (20), 83 (53), 78 (100), 77 (27), 71 (63), 70 (50), 69 (37), 57 (90), 55 (53), 43 (70), 41 (53).

2.2. Hydrogenation of 5. A solution of 5 (432 mg, 175 mmol) in AcOEt (70 ml) was hydrogenated at 3–5 atm. (Parr-Shaker) with 30 mg of 10% Pd/BaSO₄ for 6 h. The catalyst was removed by filtration. Chromatography (50 g of silica gel, cyclohexane/AcOEt 1:1) of the residue yielded 53 mg (ca. 12%) of impure 5 and 307 mg (70%) of a mixture of 18 and 19. Part (154 mg) of this mixture was separated by crystallization (CH₂Cl₂/hexane) and chromatography of the mother liquor (silica gel, hexane/CH₂Cl₂/AcOEt 1:1:1) into 103 mg (67%) of 18, 22 mg (14%) of 19, and 18 mg (ca. 12%) of a 9:1 mixture of 19 and 18.

 $(1 \text{R}^*, 4a \text{S}^*, 9b \text{R}^*)$ -7.8-Dimethoxy-1,2,3,4,4a,9b-hexahydrodibenzofuran-1-ol (18): m.p. 148–149° (CH₂Cl₂/hexane). IR (CHCl₃): 3615w, 3580w, 3035w, 2940s, 2915m, 2875m, 2840w, 1618m, 1493s, 1467s, 1455s, 1444s, 1417w, 1327w, 1360m, 1341m, 1328w, 1300m, 1292m, 1272w, 1177s, 1166s, 1112s, 1101m, 1060m, 1030m, 1000m, 978m, 927w, 906w, 897w, 887w, 875w. ¹H-NMR (100 MHz, CDCl₃): 1.2–2.1 (m, 2H–C(2), 2H–C(3), 2H–C(4), HO–C(1)); 3.41 (dd, J = 8 and 5, H–C(9b)); 3.80 (s, CH₃O–C(7), CH₃O–C(8)); 3.9–4.2 (m, H–C(1)); 4.6–4.9 (m, H–C(4a)); 6.47 (m, $w_{V_2} \approx 1$) and 6.93 (m, $w_{V_2} \approx 2$) (H–C(6), H–C(9)). MS(di.): 250 (100, M^+), 235 (13), 217 (11), 191 (65), 167 (12), 166 (14), 43 (15).

 $(1 \text{ R}, 4a \text{ S}, 9b \text{ R}^*)$ -7.8-Dimethoxy-1,2,3,4,4a,9b-hexahydrodibenzofur-1-yl Acetate (20). Acetylation of 18 (23 mg) with Ac₂O/pyridine gave 27 mg (98%) of 20, purified by chromatography (silica gel, hexane/CH₂Cl₂/Et₂O 2:2:1). IR (CHCl₃): 3030w, 2995w, 2940m, 2865w, 2830w, 1722s, 1613m, 1490s, 1461m, 1450m, 1440m, 1412w, 1375w, 1367m, 1360m, 1340w, 1325w, 1300w, 1185s, 1172m, 1161m, 1144m, 1120w, 1110m, 1098m, 1063m, 1050w, 1030m, 990m, 972m, 934w, 917w, 908w, 892w, 969w, 850m, 822w. ¹H-NMR (300 MHz, CDCl₃): 1.39–1.53 (m, 1H); 1.55–1.95 (m, 5H); 1.91 (s, CH₃COO-C(1)); 3.62 (dd, J = 8 and 5,5 H-C(9b)); 3.82 and 3.83 (2s, CH₃O-C(7), CH₃O-C(8)); 4.80 (dt, J = 8 and 5.5, H-C(4a)); 5.25 (ddd, J = 8.5, 5.5, and 4, H-C(1)); 6.46 (s, H-C(6)); 6.87 (d, $J \approx 0.7$, H-C(9)). MS(di.): 292 (100, M^+), 277 (3), 249 (10), 233 (10), 232 (55), 217 (30), 205 (17), 204 (50), 191 (63), 178 (10), 167 (13), 151 (12), 73 (27), 61 (34), 45 (74), 43 (67).

 $(1S^*,4aS^*,9bR^*)$ -7,8-Dimethoxy-1,2,3,4,4a,9b-hexahydrodibenzofuran-1-ol (19). M.p. 116–117° (CH₂Cl₂/hexane). IR (CHCl₃): 3600w, 3550–3300w, 3030w, 2995w, 2935m, 2865w, 2830w, 1614m, 1489s, 1460m, 1450m, 1440m, 1413w, 1364w, 1336w, 1298w, 1183s, 1161s, 1108s, 1080w, 1045w, 1035w, 995m, 972m, 882m, 865w, 850m, 825w. ¹H-NMR (300 MHz, CDCl₃): 1.25–1.4 (m, 1H); 1.5–1.9 (m, 4H); ≈ 1.75 (br., exchangeable with D_2O , HO–C(1)); 2.05–2.2 (m, 1H); 2.80 (dd, J = 8 and 6.5, H–C(9b)); 3.47 (m, after exchange with D_2O , ddd, J = 10.5, 8, and 4, H–C(1)); 3.827 and 3.833 (2s, CH₃O–C(7), CH₃O–C(8)); 4.73 (dt, $J \approx 6.5$ and 3.5, further broadened by small couplings, H–C (4a)); 6.49 and 6.93 (2s, H–C(6), H–C(9)). MS(di.): 250 (100, M^+), 235 (9), 232 (1), 217 (7), 191 (58), 163 (7), 147 (6), 91 (5), 77 (7), 69 (9).

2.3. 3-Methoxy-2-(2',4',5'-trimethoxyphenyl)-2-cyclohexen-1-one (22). A mixture of 5 (118 mg, 0.482 mmol) and 4 ml of a solution of NaOH (6 g) in H₂O (10 ml) and EtOH (50 ml) was boiled under reflux for 4 h (Ar). After the addition of CH₃I (1 ml and 1 ml after 14 h, 5.56 g, *ca.* 32 mmol), the solution was warmed in an oil bath (50°) for 18 h (Ar). Workup with AcOEt gave 71 mg of products (mainly 5 and 22) from the org. phases and 93 mg of 23/24 from extraction (AcOEt) of the acidified aq. phases. The partially methylated material (93 mg of 23/24) was treated with CH₃I (1 ml, 2.28 g, *ca.* 16 mmol)/K₂CO₃ (1.5 g) in acetone (5 ml) for 39 h under reflux (50°, Ar). Workup with AcOEt and chromatography (15 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 2:5:3 + 5% CH₃OH) of the combined crude products (155 mg) gave 12 mg (10%, after re-chromatography, silica gel, CH₂Cl₂/AcOEt 10:1) of 5 and 95 mg (67%, 74% based on consumed 5) of 22; m.p. 116–117° (CH₂Cl₂/hexane). IR (CHCl₃): 3030w, 2995m, 2940m, 2910w, 1845w, 2830w, 1642s, 1599s, 1505s, 1459s, 1432w, 1390m, 1362m, 1353s, 1325w, 1293m, 1275w, 1253m, 1162m, 1129m, 1079m, 1032s, 1000w, 950w, 856m, 838w. ¹H-NMR (80 MHz, CDCl₃): 1.85–2.3 (m, 2H–C(5)); 2.3–2.85 (m, 2H–C(4), 2H–C(6)); 3.64, 3.70, 3.79, and 3.85 (4s, 4 OCH₃); 6.54 (s, H–C(3'), H–C(6')). MS (*di.*): 292 (100, M^+), 277 (28), 261 (26), 249 (17), 246 (7), 191 (9), 175 (5), 163 (5), 161 (5), 147 (5), 131 (6), 115 (6), 103 (10), 91 (10), 77 (11), 69 (17), 55 (24). Anal. calc. for C₁₆H₂₀O₅ (292.32): C 65.74, H 6.90; found: C 65.63, H 6.84.

2.4. Reduction of 12 with NaBH₄. To a solution of 12 (229 mg, 0.835 mmol) in CH₃OH (50 ml) a solution of NaBH₄ (635 mg, 16.7 mmol) in H₂O (4 ml) was added in portions. After stirring for 2 h at r.t., the mixture was worked up with AcOEt giving 190 mg (97%) of crude 7-methoxy-1,2,3,4-tetrahydrodibenzofuran-1,8-diol (25): IR (CHCl₃): 3600w, 3545m, 2945m, 2870w, 2845w, 1628m, 1598w, 1476s, 1460m, 1440s, 1370s, 1358m, 1335m, 1315m, 1272m, 1255m, 1190m, 1161w, 1137s, 1121m, 1056w, 1023w, 981w, 961w, 954w, 930w, 893w, 884m, 860m, 823w.

7-Methoxy-1,2,3,4-tetrahydrodibenzofuran-1,8-diyl Diacetate (26). Crude 25 (106 mg) was acetylated with Ac₂O/pyridine (0.75 ml of each) giving 142 mg (95% based on 12) of 26. IR (CHCl₃): 3035w, 2955m, 2945m, 2875w, 2850w, 1757s, 1722s, 1630m, 1592w, 1488s, 1452m, 1438m, 1372s, 1350w, 1330m, 1316m, 1275m, 1263s, 1180m, 1160m, 1146s, 1120m, 1078w, 1053w, 1027m, 1014m, 977w, 962m, 950w, 941w, 920m, 904m, 892w, 870w. ¹H-NMR (100 MHz, CDCl₃): 1.84–2.12 (m, 2H–C(2), 2H–C(3)); 2.03 (s, CH₃COO–C(1)); 2.29 (s, CH₃COO–C(8)); 2.58–2.84 (m, 2H–C(4)); 3.80 (s, CH₃O–C(7)); 6.00–6.16 (m, H–C(1)); 7.00 and 7.19 (2s, H–C(6), H–C(9)). MS (di.): 318 (33, M^+), 276 (34), 259 (14), 258 (16), 234 (100), 217 (40), 216 (77), 201 (34), 43 (48).

2.5. Ethyl (8'-Benzyloxy-7'-methoxy-1',2',3',4'-tetrahydrodibenzofuran-1'-ylidene) acetate (27). To a suspension of LiNH₂ in NH₃ (liq.), prepared by FeCl₂-catalyzed reaction of Li (658 mg, 93 mgAt) with NH₃ (ca. 20 ml, distilled from Na), 2-bromovinylethylether (7.753 g, 51.4 mmol) was added at -78° . After stirring for 1 h at -78° , dry THF (100 ml) was added and most of the NH₃ was evaporated. This reagent solution was cooled to -78° , and 14 (1.5 g, 4.66 mmol) dissolved in THF (10 ml) was slowly added. After stirring for 15 min at -78° , the mixture was warmed to r.t., quenched by the careful addition of sat. (NH₄)₂SO₄-soln., acidified (2N HCl) and worked up with AcOEt. Flash chromatography (100 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 12:24:1) afforded 1.374 g (75%, 87% based on consumed 14) of 27 and 214 mg (14%) of starting mat. 14. M.p. 138-139°. IR (CHCl₃): 3095w, 3060w, 3035w, 2985w, 2940m, 2905w, 2875w, 2842w, 1695s, 1620s, 1488s, 1451s, 1373w, 1348s, 1296s, 1253w, 1167s, 1120s, 1030m, 1010m, 960w, 943w, 879w, 864w. ¹H-NMR (300 MHz, $CDCl_3$: 1.36 (t, J = 7, $CH_3CH_2O-C(1)$); 2.01 (quint., $J \approx 6.5$, 2H-C(3')); 2.83 (t, $J \approx 6.5$, 2H-C(4')); 3.13-3.22 (m, 3 main peaks, 2H-C(2')); 3.91 (s, CH₃O-C(7')); 4.23 (q, CH₃CH₂O-C(1)); 5.18 (s, C₆H₅CH₂O-C(8)); 6.10 (m, $w_{1/2} \approx 4$, H-C(2)); 7.02 (s, H-C(6')); 7.25-7.42 (m, 3H) and 7.48-7.55 (m, 2 main peaks, H-ortho) $(C_6H_5CH_2O-C(8'));$ 7.29 (s, H-C(9')). MS (di.): 392 (59, M⁺), 347 (8), 302 (21), 301 (100), 273 (7), 257 (19), 227 (14), 200 (9), 199 (11), 171 (9), 128 (9), 115 (7), 91 (46). Anal. calc. for $C_{24}H_{24}O_5$ (392.43): C 73.45, H 6.16; found C 73.49, H 6.33.

2.6. Hydrogenation of 27. A solution of 27 (207 mg, 0.528 mmol) in dry EtOH was hydrogenated at normal pressure with 5% Pd/C (*ca.* 5 mg). After 15 h of reaction, TLC analysis (cyclohexane/CH₂Cl₂/AcOEt 6:12:1) showed, that only partial hydrogenation had occurred. Addition of more 5% Pd/C (*ca.* 10 mg) was followed by

stirring under H₂ for 36 h. Removal of catalyst by filtration (*Celite*) and evaporation of solvent afforded 142 mg of crude **28**, 74 mg of which was acetylated (Ac₂O/pyridine). Chromatography (8 g of silica gel, cyclohexane/ CH₂Cl₂/AcOEt 6:12:1) gave 2.5 mg (*ca.* 2.5%) of **33** and 51 mg (53%) of **29**.

Ethyl [(1'S*, 4a'S*, 9b'R*)-8'-Acetoxy-7'-methoxy-1', 2', 3', 4', 4a', 9b'-hexahydrodibenzofuran-1'-yl]acetate (29). IR (CHCl₃): 3030w, 2980m, 2940s, 2860w, 2835w, 1755s, 1724s, 1625s, 1600w, 1493s, 1466m, 1450s, 1422m, 1370s, 1347m, 1327m, 1313m, 1300m, 1266m, 1180s, 1169s, 1112m, 1097m, 1060m, 1045m, 1025m, 1018m, 1003m, 980m, 920m, 885w. ¹H-NMR (100 MHz, CDCl₃): 1.24 (t, J = 7, CH₃CH₂O); 1.06–1.96 and 2.14–2.52 (2m, 9H); 2.24 (s, CH₃COO-C(8')); 3.49 (dd, $J \approx 8$ and 4, H–C(9b')); 3.73 (s, CH₃O-C(7')); 4.12 (q, J = 7, CH₃CH₂O); 4.90 (dt, $J \approx 8$ and 4, H–C(4a')); 6.42 (s, H–C(6')); 6.86 (m, $w_{V_2} \approx 3$, H–C(9')). MS: 348 (15, M^+), 307 (20), 306 (100), 261 (10), 260 (8), 218 (23), 217 (9), 177 (23), 43 (10).

Ethyl (8'-Acetoxy-7'-methoxy-1',2',3',4'-tetrahydrodibenzofuran-1'-yl)acetate (33). IR (CHCl₃): 3035w, 2930m, 2860w, 2840w, 1755s, 1723s, 1625w, 1480m, 1447m, 1435m, 1368m, 1335m, 1325m, 1316m, 1276m, 1260m, 1165s, 1140s, 1115m, 1095w, 1067w, 1021m, 960w, 916m, 902w, 890w, 865w. ¹H-NMR (100 MHz, CDCl₃): 1.23 (t, J = 7, CH₃CH₂O); 1.5–2.1 (m, 2H–C(2'), 2H–C(3')); 2.28 (s, CH₃COO–C(8')); 2.35 (dd, J = 13 and 10) and 2.81 (dd, J = 13 and 5) (2H–C(2)); 2.55–2.80 (m, 2H–C(4')); 3.2–3.5 (m, H–C(1')); 3.80 (s, CH₃O–C(7')); 4.13 (q, J = 7, CH₃CH₂O); 7.00 and 7.08 (2s, H–C(6'), H–C(9')). MS: 346 (35, M ⁺), 304 (100), 259 (11), 217 (50), 216 (35), 205 (12), 201 (8), 91 (9), 85 (34), 55 (14), 43 (21).

2.7. Oxidation of **28** with HIO₄. To a solution of HIO₄·2H₂O (60 mg, 0.263 mmol) in H₂O (2.5 ml), a solution of crude **28** (68 mg, 0.222 mmol) in CH₃CN (10 ml) was added at 0°. After 1 min the orange colored mixture was worked up with Et₂O. The residue of the org. layers (65 mg) was dissolved in CH₂Cl₂ and shaken with an aq. solution of NaS₂O₄. Workup with CH₂Cl₂, acetylation (Ac₂O/pyridine), and chromatography (13 g of silica gel, cyclohexane/CH₂Cl₂/AcOEt 6:12:1) of the crude mixture (83 mg) gave 8 mg (10%) of **33** and 20 mg (*ca.* 24%) of impure *ethyl* (7',8'-*diacetoxy-1',2',3',4'-tetrahydrodibenzofuran-1-yl)acetate* (**34**). IR (CHCl₃): 3030w, 2936m, 2935m, 2865w, 2845w, 1767s, 1725s, 1623w, 1493w, 1472s, 1448s, 1435m, 1370s, 1345m, 1326m, 1306m, 1285m, 1165s, 1129s, 1107s, 1080m, 1069m, 1030m, 1010m, 970w, 940w, 920m, 903m, 893m, 865w. 'H-NMR (90 MHz, CDCl₃): 1.23 (*t*, *J* = 7, CH₃CH₂O₂); 1.3–2.1 (*m*, 2H–C(2'), 2H–C(3')); 2.27 (*s*, CH₃COO–C(7'), CH₃COO–C(8')); 2.35 (*dd*, *J* = 15 and 10) and 2.80 (*dd*, *J* = 15 and 5) (2H–C(2)); 2.5–2.8 (*m*, 2H–C(4')); 3.1–3.55 (*m*, H–C(1')); 4.13 (*q*, CH₃CH₂O); 7.24 (*m*, $w_{12} \approx 2$, H–C(6'), H–C(9')).

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